

Date Planned : __ / __ / __	Daily Tutorial Sheet-6	Expected Duration : 45 Min
Actual Date of Attempt : __ / __ / __	JEE Advanced (Archive)	Exact Duration : _____

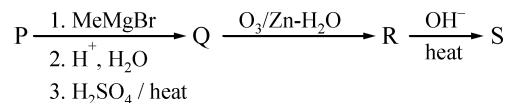
75. Match each of the compounds given in Column I with the reaction(s) that they can undergo, given in Column II. (2008)

Column I		Column II	
(A)		(p)	Nucleophilic substitution
(B)		(q)	Elimination
(C)		(r)	Nucleophilic addition
(D)		(s)	Esterification with acetic anhydride
		(t)	Dehydrogenation

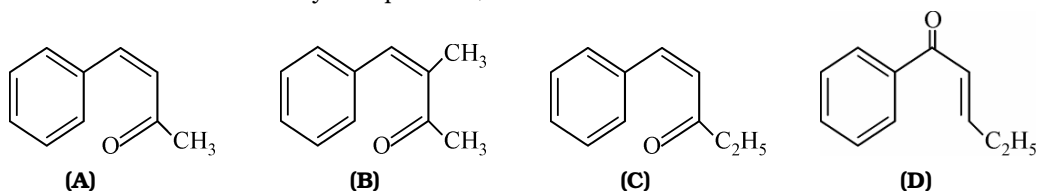
76. Which of the following on heating with aqueous KOH produced acetaldehyde? (2009)  
(A)  $\text{CH}_3\text{COCl}$  (B)  $\text{CH}_3\text{CH}_2\text{Cl}$  (C)  $\text{CH}_2\text{ClCH}_2\text{Cl}$  (D)  $\text{CH}_3\text{CHCl}_2$

**PARAGRAPH FOR QUESTIONS 77 - 79**

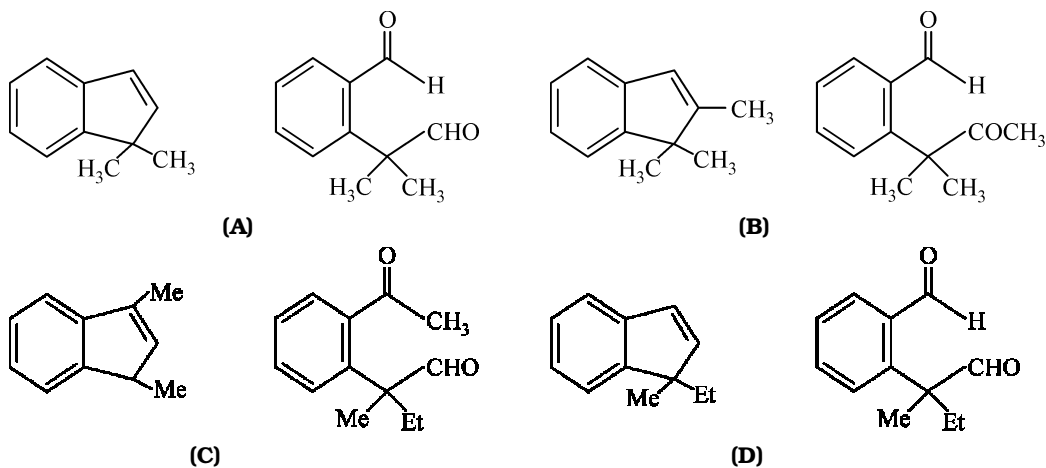
A carbonyl compound P, which gives positive iodoform test, undergoes reaction with  $\text{MeMgBr}$  followed by dehydration to give an olefin Q. Ozonolysis of Q leads to a dicarbonyl compound R, which undergoes intramolecular aldol reaction to give predominantly S. (2009)



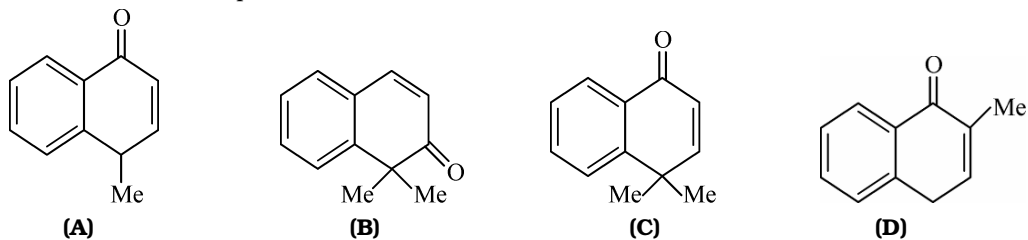
77. The structure of the carbonyl compound P, is :



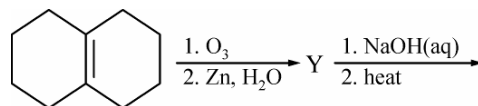
78. The structure of the products Q and R, respectively, are :



79. The structure of the product S, is :

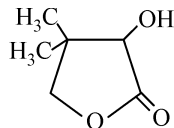


80. In the scheme given below, the total number of intramolecular aldol condensation products formed from Y is : (2010)



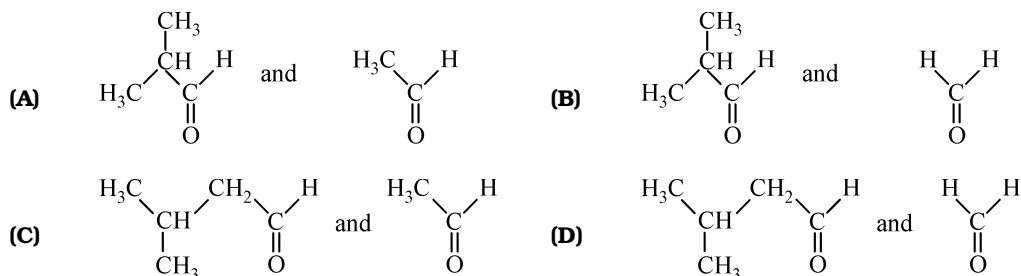
**PARAGRAPH FOR QUESTIONS 81 - 83**

Two aliphatic aldehydes P and Q react in the presence of aqueous  $K_2CO_3$  to give compound R, which upon treatment with HCN provides compound S. On acidification and heating, S gives the product shown below :



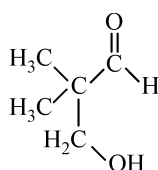
81. The compounds P and Q respectively are :

(2010)

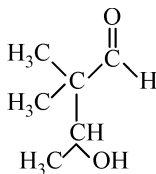


82. The compound R is :

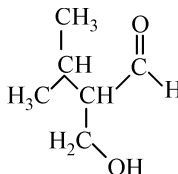
(2010)



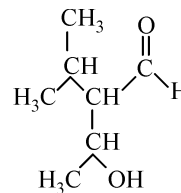
(A)



(B)



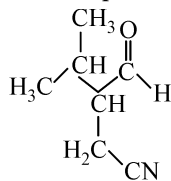
(C)



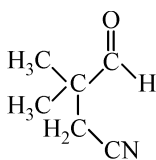
(D)

83. The compound S is :

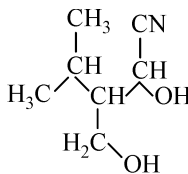
(2010)



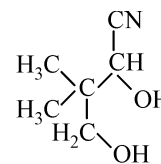
(A)



(B)



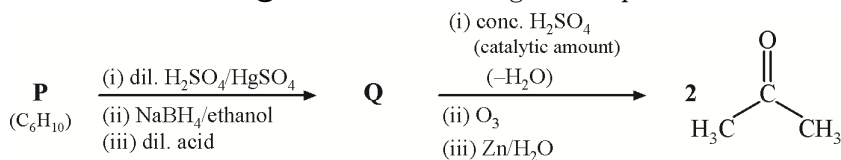
(C)



(D)

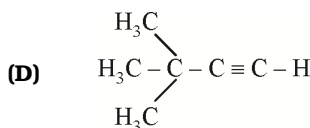
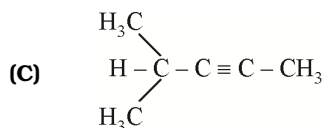
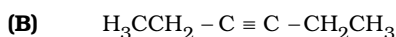
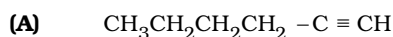
**PARAGRAPH FOR QUESTIONS 84 - 85**

An acyclic hydrocarbon **P**, having molecular formula  $C_6H_{10}$  gave acetone as the only organic product through the following sequence of reactions, in which **Q** is an intermediate organic compound.



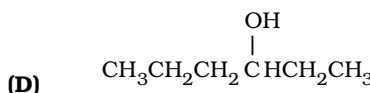
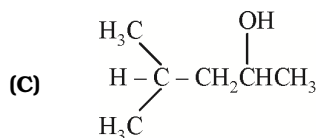
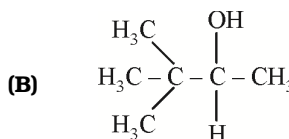
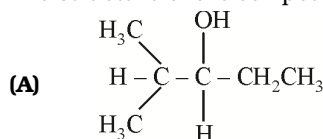
84. The structure of compound **P** is :

(2011)



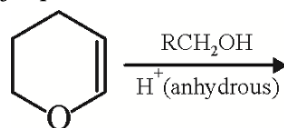
85. The structure of the compound **Q** is :

(2011)



86. The major product of the following reaction is :

(2011)



- (A) A hemiacetal    (B) An acetal    (C) An ether    (D) An ester

87. Match the reactions in Column I with appropriate types of steps/reactive intermediate involved in these reactions as given in Column II. (2011)

Column I		Column - II	
(A)		(p)	Nucleophilic substitution
(B)		(q)	Electrophilic substitution
(C)		(r)	Dehydration
(D)		(s)	Nucleophilic addition
		(t)	Carbanion

88. The compound that undergoes decarboxylation most readily under mild condition is: (2012)

